

Birth and life of sulphur dimers (S₂) in cometary ices

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S₂ has been observed for decades in comets, but its recent observation in comet 67P/Churyumov-Gerasimenko (le Roy et al 2015) has reactivated new interest. As a matter of fact, the nature of its source is still in debate. Relying upon our scenario for O₂, we propose that S₂ is formed according an analogous process, i.e. by irradiation (photolysis and/or radiolysis) of S-bearing molecules embedded in the icy grains precursors of comets. As O₂ could be issued from the fragments of H₂O produced by irradiation, S₂ could come from the fragmentation of H₂S. This precursor is known as the most abundant sulfur-containing molecule present in comets ices, and could exist scattered or as clumps in the bulk of H₂O ices. The irradiation is assumed to create simultaneously voids in ices within which the produced molecules can accumulate. We investigate the stability of S₂ molecules in such cavities, considering that the surrounding ice is made whether of H₂S or of H₂O. We have studied this scenario by using quantum chemistry numerical models based on first principle periodic density functional theory (DFT). Such models have proved to be well adapted to the description of compact ice and are capable at the same time to describe the trapping of volatiles in cavities inside the ice matrix (Ellinger et al 2015, Mousis et al 2016). We show that the stabilization energy of S₂ molecules in such voids is close to that of the H₂O ice binding energy, which implies that they can only leave the icy matrix when this latter sublimates. Unlike O₂ whose abundance correlated to H₂O, no global trend should be drawn between the variation of S₂ and H₂O abundances, which can be explained by the fact that S₂ can accumulate in both S-bearing and H₂O ices. This interpretation is supported by the ROSINA data collected between May 2015 (equinox) and August 2015 (perihelion), showing that there is no clear correlation of S₂ with H₂O or H₂S in 67P/C-G (Calmonte et al. 2016).